ARBUTIN IN SOME PLANTS OF THE GENUS SEDUM

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In a study of the composition of three species of <u>Sedum</u>: <u>S. hybridum</u> L., <u>S. aizoon</u> L., and <u>S. purpureum</u> Link (Crassulaceae) by chromatography in a nonfixed layer of alumina in a butan-1-ol-ethanol-water (5:1:2) system, we found a substance of phenolic nature with $R_f 0.41$ (detection with diazotized sulfanilic acid).

In order to isolate the substance, the epigeal parts of the plants were extracted with ethanol and the dry residue was separated on a column of alumina (activity grade II, neutral) by discrete-gradient elution with chloroformethanol systems ($10: 0 \rightarrow 1: 9$). When the column was eluted with the systems having ratios of 1:1 and 3:7, a phenolic glycoside was eluted in the form of colorless acicular crystals, $C_{12}H_{16}O_7 \cdot H_2O$, mp 161–162° C (from a mixture of chloroform and ethanol), $[\alpha]_D^{20}$ -64.7° (c 0.91, water). Its UV spectrum has maxima at 220 and 283 m μ (log ϵ 2.84 and 2.33, respectively). After drying in a vacuum pistol over P_2O_5 at 110° C for 3 hr, a hygroscopic anhydrous glycoside with mp 195–195.5° C was obtained. The substance forms a pentaacetate with mp 145–145.5° C.

On hydrolysis of the glycoside with 5% H_2SO_4 and with the enzymes of the fungus <u>Aspergillus oryzae</u>, equimolecular amounts of hydroquinone (mp 168.5-169°C; melting point of the diacetate 120-121°C) and D-glucose were formed.

According to its qualitative reactions [1], cleavage products, UV and IR spectra, and Rf values, and a mixed melting point test, the isolated glycoside was identified as arbutin [2, 3].

From <u>S. aizoon</u> two substances were isolated, with mp 139-140° C and mp 298-300° C, which were identified by their chemical properties (Lieberman-Burchard reaction) and mixed-melting point tests as β -sitosterol and oleanolic acid, respectively.

This is the first time that arbutin has been isolated from the family Crassulaceae.

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